The Structure and Properties of β -La₃RuO₇: A New Structure **Type with Isolated RuO6 Octahedra**

P. Khalifah,*^{,1} D. M. Ho,* Q. Huang, \ddagger ;§ and R. J. Cava*;†

**Department of Chemistry, Princeton University, Princeton, New Jersey 08544-1009;* -*Princeton Materials Institute, Princeton University, Princeton, New Jersey 08540;* ?*NIST Center for Neutron Research, National Institute of Standards and Technology, Gaithersburg, Maryland 20899; and* A*Department of Materials and Nuclear Engineering, University of Maryland, College Park, Maryland 20742*

Received September 17, 2001; in revised form February 7, 2002; accepted February 22, 2002

A second lanthanum ruthenium oxide with the formula La₃RuO₇ is reported. This compound, β -La₃RuO₇, crystallizes in the $P2_1/c$ space group with a unit cell of $a = 8.8388(3)$ Å, $b = 5.6960(1)$ Å, $c = 12.5830(3)$ Å, and $\beta = 104.729(1)$ °, as refined from powder neutron diffraction data. The two major structural motifs in this structure are isolated $RuO₆$ octahedra and edge-sharing chains of $OLa₄$ tetrahedra. Magnetic susceptibility measurements show this compound to have a magnetic moment of $\mu_{\text{eff}} = 3.92 \mu_{\text{B}}$ and a Weiss constant of $\theta = -$ 70 K. \circ 2002 Elsevier Science (USA)

INTRODUCTION

The $Ln_3MO_7[Ln = La-Lu, Y; M = Sb (1), Ta (1), Nb (2),$ $Ln_3MO_7[Ln = La-Lu, Y; M = Sb (1), Ta (1), Nb (2),$ $Ln_3MO_7[Ln = La-Lu, Y; M = Sb (1), Ta (1), Nb (2),$ $Ln_3MO_7[Ln = La-Lu, Y; M = Sb (1), Ta (1), Nb (2),$ $Ln_3MO_7[Ln = La-Lu, Y; M = Sb (1), Ta (1), Nb (2),$ Mo [\(3\),](#page-3-0) Re [\(4\)](#page-3-0), Ru [\(5\)](#page-3-0), Rh [\(6\)\]](#page-3-0) family of quasi-1D materials contains corner-sharing chains of $MO₆$ octahedra, making them good candidate materials for the study of low-dimensional magnetism. The magnetic properties of this family are also interesting because this structure type exists for a wide range of *d*-electron counts, from d^0 to d^4 , giving rise to chains with local moments that range from $S = 0$ through $\frac{3}{2}$. The subset of compounds with $M = \text{Ru} (d^3)$ are the only members of this family to have the maximum transition metal low-spin local moment of $S = \frac{3}{2}$. Among the subset of d^3 Ln_3RuO_7 compounds, α -La₃RuO₇ is the only known compound with a nonmagnetic rare earth, providing a chance to study the physics of a $S = \frac{3}{2}$ chain within a nonmagnetic matrix.

Unfortunately, α -La₃RuO₇ is the most challenging member of the known Ln_3RuO_7 compounds to synthesize. Although this compound was originally reported to be impossible to synthesize [\(6, 7\),](#page-3-0) later work showed that it is possible to make this compound using a precursor lanthanum ruthenate with an unknown composition and structure [\(8, 9\)](#page-3-0). Here, we report the crystal structure and magnetic behavior of this precursor compound, β - $La_3RuO_7.$

Although the initial interest in β -La₃RuO₇ involved its value as a synthetic precursor for making α -La₃RuO₇, the antiferromagnetic behavior of β -La₃RuO₇ provides a valuable opportunity for understanding the magnetism of highly oxidized Ru compounds with isolated octahedra. Two other lanthanum ruthenates are known to have isolated RuO_6 octahedra, $La_{4.87}Ru_{2}O_{12}$ and $La_{7}Ru_{3}O_{18}$ [\(10\)](#page-3-0). However, their $RuO₆$ octahedra are arranged in a geometrical array conducive to geometric frustration. The measured magnetic behavior of β -La₃RuO₇ provides a baseline for understanding the magnetism of highly oxidized Ru states in the absence of complicating geometric frustration, allowing the magnetism of the frustrated compounds $La_{4.87}Ru_2O_{12}$ and $\text{La}_7\text{Ru}_3\text{O}_{18}$ to be put in perspective.

EXPERIMENTAL

Powder samples were prepared by mixing dried $La₂O₃$ and $RuO₂$ in a 3.33:1 ratio of La:Ru. The well-ground mixture was placed in a dense Al_2O_3 and fired in air at $10000G_4$ 1000° C for 1 day and was further heated in air at 1100 $^{\circ}$ C for few days with multiple regrindings. The reaction was judged complete when there were no longer any $La_{4.87}Ru_2O_{12}$
X-ray diffraction peaks [\(10\)](#page-3-0) present. It was found that in air, 1100°C was the highest temperature that β -La₃RuO₇ could be prepared without decomposing into other lanthanum ruthenates, while the synthesis of α -La₃RuO₇ required a temperature of 1250 \degree C and 1 atm of O_2 to prevent the formation of other lanthanum ruthenates. At intermediate temperatures, a mixture of the two phases was present. Although tiny single crystals of β -La₃RuO₇ could be obtained from crystal growths in salt fluxes (NaCl, KCl), these crystals diffracted too weakly to allow an accurate structure determination to be made. Nonetheless, unit cell and space group information $(P2_1/c)$, and the approximate atomic

¹ To whom correspondence should be addressed at Solid State Division, Oak Ridge National Laboratory, P.O. Box 2008, Oak Ridge, TN 37831- 6056.

parameters could be obtained using long exposure times (120 s/deg). These approximate values were used as the initial input for neutron diffraction refinements. Both fieldcooled and zero-field-cooled magnetic susceptibility measurements were performed on powder samples of β - $La₃RuO₇$ using a Physical Property Measurement System (Quantum Design).

Powder neutron diffraction data were collected at the NIST Center for Neutron Research using the BT-1 32 counter high-resolution powder diffractometer. Room temperature data were collected using neutrons of wavelength 1.5401 A produced by a Cu (311) monochromator. Collimators with horizontal divergences of $15'$, $20'$, and $7'$ of arc full-width at half-maximum were used before and after the monochromator, and after the samples, respectively. Intensities were measured in steps of 0.05 $^{\circ}$ in the 2 θ range 3–168 $^{\circ}$. The lattice parameters were determined using GSAS [\(11\)](#page-3-0) on data collected at 298 K. The neutron scattering amplitudes used in all calculations were $b(La) = 0.827$, $b(Ru) = 0.721$, and $b(O) = 0.581$ ($\times 10^{-12}$ cm).

RESULTS AND DISCUSSION

Neutron diffraction data (collected on a \sim 10 g powder sample) were used to refine the structure of β -La₃RuO₇ in the monoclinic space group $P2_1/c$ (no. 14) with a unit cell of 8.8388(2) $\text{\AA} \times 5.6960(1) \text{\AA} \times 12.5830(3) \text{\AA}$ and $\beta = 104.729(1)^\circ$. Structural parameters at room temperature are listed in Tables 1 and 2. The final three-phase refinement took into account the presence of two impurity phases (7.8 wt) La₂O₃ and 3.2 wt% α -La₃RuO₇) in addition to the majority phase (89.0 wt% β -La₃RuO₇). The calculated χ^2 (0.985) refined to a value less than 1 due to the underestimation of experimental errors, a problem that may occur when samples smaller than the instrumental norm (\sim 20 g) are studied. The observed intensities, calculated intensities, and their difference are shown in Fig. 1.

An examination of the structure of β -La₃RuO₇ shows the two major structural motifs to be isolated $RuO₆$ octahedra

TABLE 2 Atomic Coordinates and Thermal Parameters

Atom	Wyck.	X	Y	Z	$100U_{\rm iso}$
La1	4e	$-0.0022(2)$	0.0217(4)	0.1400(1)	0.75
La2	4e	0.2970(2)	0.0180(4)	0.4273(1)	0.67
La ₃	4e	0.4307(2)	$-0.0089(4)$	0.1331(1)	0.60
Ru	4e	$-0.2143(3)$	0.0434(4)	0.3333(2)	0.83
O ₁	4e	$-0.2511(4)$	$-0.1256(5)$	0.1969(3)	1.35
O ₂	4e	0.0079(4)	0.0567(5)	0.3328(2)	1.16
O ₃	4e	$-0.1700(4)$	0.2654(6)	0.4658(2)	0.71
O ₄	4e	$-0.4328(3)$	0.0440(5)	0.3262(2)	0.74
O ₅	4e	$-0.2385(4)$	0.3416(5)	0.2528(3)	1.18
O ₆	4e	$-0.1891(4)$	$-0.2410(5)$	0.4273(2)	1.04
O ₇	4e	0.4917(4)	$-0.2625(5)$	0.4897(2)	0.66

and chains of edge-sharing $OLa₄$ tetrahedra. The connectivity of these chains, which run parallel to the *b*-axis, is readily seen when β -La₃RuO₇ is viewed along the *a*-axis [\(Fig. 2\).](#page-2-0) The separation of these structural motifs is best seen in views along the *b*-axis [\(Fig. 3\)](#page-2-0). Two-thirds of the La atoms $(La2 \text{ and } La3)$ are found in $OLa₄$ tetrahedra, while the other third of the La atoms (La1) only coordinate oxygens that are part of $RuO₆$ octahedra. Taking a cutoff of 3.2 Å as the upper limit for the $La-O$ bond length, we find that all three La sites are nine-coordinate, and that all three La coordination polyhedra are quite irregular. The La–O bond distances in this structure are given in [Table 3.](#page-2-0) It is interesting to note that the four shortest $La-O$ bonds in the entire structure are found in the $OLa₄$ tetrahedron. The $OLa₄$ tetrahedra are quite regular, as all angles are within 10 \degree of the 109.5 \degree expected for an ideal tetrahedron. The isolated RuO₆ octahedra have no abnormal distortions. The O-Ru-O

FIG. 1. Observed and calculated powder neutron diffraction pattern for β -La₃RuO₇ (89.0 wt%), La₂O₃ (7.8 wt%), and α -La₃RuO₇ (3.2 wt%) at room temperature. Below the observed data (crosses) and calculated (solid line) pattern, the difference plot is shown. Vertical lines indicate the Bragg reflection positions for β -La₃RuO₇ (bottom), La₂O₃ (middle), and α - La_3RuO_7 (top).

FIG. 2. Structure of β -La₃RuO₇ viewed nearly parallel to the *a*-axis, showing the isolated $RuO₆$ octahedra and the edge-sharing chains of OLa₄ tetrahedra. The $RuO₆$ octahedra are shown in light gray while OLa₄ tetrahedra are shown in dark gray. Oxygen atoms are represented a black spheres, while La atoms are shown as large gray spheres. Ru atoms (medium spheres) are only found at the center of $RuO₆$ octahedra.

bond angles fall between 86 and 95° , and the Ru-O bond lengths ranging from 1.91 to 2.05 Å.

We briefly note that there is no simple distortion that will effect the transformation between β -La₃RuO₇ and α - $La₃RuO₇$. This is perhaps best seen in the two different connectivities of the OLa₄ tetrahedral networks. In addition to having the same stoichiometry and the same $Ru⁵⁺$ oxidation state, both compounds also have four formula units per unit cell. The smaller cell volume of β -La₃RuO₇ (612.7 vs) 636.3 Å³ for α -La₃RuO₇) means that this low-temperature structure type is about 4% denser than α -La₃RuO₇. The previously observed transformation from β -La₃RuO₇ to α -La₃RuO₇ on heating [\(8, 9\)](#page-3-0) is believed to be more a result of chemical similarity than structural similarity.

In the two other lanthanum ruthenates with isolated RuO_6 octahedra, $La_{4.87}Ru_2O_{12}$ and $La_7Ru_3O_{18}$ [\(10\),](#page-3-0) the $RuO₆$ octahedra are arranged around perfect or near-perfect three-fold rotation axes, leading to geometrical frustration of antiferromagnetic ordering. However, the spatial arrangement of octahedra in β -La₃RuO₇ seems to be very irregular. This absence of geometric order may allow β - $La₃RuO₇$ to serve as a benchmark for the unfrustrated magnetism of ruthenates.

A plot of the inverse magnetic susceptibility of β -La₃RuO₇ (*H* = 1.0 T) shows that Curie–Weiss behavior persists down to about 40 K [\(Fig. 4\).](#page-3-0) A magnetic moment of $\mu = 3.92 \mu_B$ and a Weiss constant of $\theta = -70$ K are obtained from the fit parameters after correcting for the

 La_3RuO_7 viewed parallel to the *b*-axis. The separation between the $OLa₄$ chains (which run parallel to *b*) can be clearly observed. It can also be seen that one axis of the $RuO₆$ octahedra is nearly parallel to the *a*-axis. Atomic and polyhedral representations are the same as in Fig. 2.

presence of 7.8 wt% nonmagnetic impurities (La_2O_3) . No correction was made for the presence of 3.2 wt\% α - $\text{La}_3 \text{RuO}_7$ due to its small volume fraction and its nearly identical magnetic moment. The moment of β -La₃RuO₇ is very close to the ideal moment of $3.87 \mu_B$ for Ru^{5+} ions $(S = \frac{3}{2})$. As expected, the Weiss constant is very similar to the values found for $La_{4.87}Ru_2O_{12}$ (-85 K) and $La_7Ru_3O_{18}$ (-58 K), which also have isolated RuO_6

TABLE 3 Selected Bond Distances (A)

$Ru-O4$	1.910(4)	$La1-O6$	2.476(4)	$La2-O3$	2.539(4)	
$Ru-O1$	1.922(4)	$La1-O3$	2.611(4)	$La2-O6$	2.600(4)	
$Ru-O5$	1.962(4)	$La1-O1$	2.619(4)	$La2-O4$	2.981(3)	
$Ru-O2$	1.967(4)	$La1-O2$	2.673(4)	$La2-O4$	3.049(3)	
$Ru-O6$	1.984(4)	$La1-O3$	2.692(4)			
$Ru-O3$	2.049(4)	$La1-O2$	3.068(4)	$La3-O7$	2.317(4)	
		La1 $-$ O6	3.194(4)	$La3-O7$	2.395(4)	
$O7 - La3$	2.317(4)			$La3-O4$	2.442(3)	
O7–La2	2.333(4)	$La2-O7$	2.333(4)	$La3-O6$	2.576(4)	
$O7 - La2$	2.351(4)	$La2-O7$	2.351(4)	$La3-O4$	2.597(4)	
$O7-I.a3$	2.395(4)	$La2-O5$	2.411(4)	$La3-O5$	2.630(4)	
		$La2-O1$	2.531(4)	$La3-O3$	2.655(4)	
$La1-O2$	2.414(3)	$La2-O1$	2.539(3)	$La3-O1$	2.802(4)	
$La1-O5$	2.437(4)			$La3-O4$	3.190(4)	

0.020

0.016

 0.012

 0.008

0.004

 Ω

 χ (emu/mol Ru)

FIG. 4. Inverse magnetic susceptibility of β -La₃RuO₇ at *H* = 1.0 T. Solid line shows the fit to a Curie-Weiss law. Inset: zoom of the lowtemperature regime.

octahedra. Upon cooling below 18 K, β -La₃RuO₇ undergoes an antiferromagnetic ordering transition, seen as a peak in the magnetic susceptibility data (Fig. 5). The temperature of this transition is noticeably higher than the corresponding transitions in $La_{4.87}Ru_2O_{12}$ (4.5 K) and $La_7Ru_3O_{18}$ (14 K). Although $La_7Ru_3O_{18}$ appears to have a rich magnetic phase diagram, the behavior of β -La₃RuO₇ is consistent with a virtually field-independent antiferromagnetic ordering transition. We note that identical behavior was obtained with zero-field cooling using a 0.1 T field and with field cooling under a 1.0 T field.

CONCLUSIONS

The structure and magnetic properties of β -La₃ have been characterized. The stoichiometry and Ru oxidation state have been found to be identical to those of the quasi-1D material, α -La₃RuO₇, making this new material an appropriate synthetic precursor for the α phase. The strength of the magnetic interactions in β -La₃RuO₇ are observed to be comparable to those of other lanthanum ruthenates with isolated octahedra, though the antiferromagnetic ordering temperature of β -La₃RuO₇ is the highest to date among these compounds.

FIG. 5. Magnetic susceptibility of β -La₃RuO₇ at *H* = 1.0 T. Solid line shows the fit to a Curie-Weiss law. Inset: zoom of the low-temperature regime.

 0.020

0.018

0.016

 0.014

0.012

 0.010^{L}_{O}

150

 β -La₃RuO₇

100

50

aar^{aax}

20

200

T(K)

40

250

60

300

80

100

350

400

ACKNOWLEDGMENTS

This work was supported by the National Science Foundation grant DMR-9725979. P.G.K. gratefully acknowledges the support of a National Science Foundation Graduate Fellowship.

REFERENCES

- 1. H. J. Rossell, *J*. *Solid State Chem*. 27, 115 (1979).
- 2. A. Kanh-Harari, L. Mazerolles, D. Michel, and F. Robert, *J*. *Solid State Chem*. 116, 103 (1995).
- 3. J. E. Greedan, N. P. Raju, A. Wegner, P. Gougeon, and J. Padiou, *J*. *Solid State Chem*. 129, 320 (1997).
- 4. G. Wltschek, H. Paulus, I. Svoboda, H. Ehrenberg, and H. Fuess, *J*. *Solid State Chem*. 125, 1 (1996).
- 5. W. A. Groen, F. P. F. van Berkel, and D. J. W. Ijdo, *Acta*. *Crystallogr*. *Sect*. *C* 43, 2264 (1987).
- 6. J. F. Vente and D. J. W. Ijdo, *Mater*. *Res*. *Bull*. 26, 1255 (1991).
- 7. F. P. F. van Berkel and D. J. W. Ido, *Mater*. *Res*. *Bull*. 21, 1103 (1986). 8. P. Khalifah, Q. Huang, J. W. Lynn, R. W. Erwin, and R. J. Cava,
- *Mater*. *Res*. *Bull*. 35, 1 (2000). 9. P. Khalifah, Q. Huang, J. W. Lynn, R. W. Erwin, B. Batlogg, and R. J. Cava, *Phys*. *Rev*. *B* 60, 9573 (1999).
- 10. P. Khalifah, Q. Huang, D. M. Ho, H. W. Zandbergen, and R. J. Cava, *J*. *Solid State Chem*. 155, 189 (2000).
- 11. A. C. Larson and R. B. Von Dreele, "General Structure Analysis System,'' Report no. LAUR086-748, Los Alamos National Laboratory, Los Alamos, NM 87545.

